

Meeting Evolving Performance Requirements for Iron and Manganese Reduction in Groundwater

Presentation to NWWC 2023

Nov 15 2023



Common Pairings



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 Common Pairings



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Intro



Who is Magnor



Guidelines



Technology
overview



Mn < 0.02



Colloidal Iron

WHO IS MAGNOR

A background image featuring a dynamic splash of clear water. The water is captured in motion, with various droplets and ripples visible, creating a sense of freshness and movement. The overall color palette is light and airy, dominated by whites and pale blues.

 Who is Magnor



Who is Magnor

- Canadiens Fans
- Supplier
- From Quebec

But we can still do good things



Company Profile

- Magnor specializes in the design and manufacturing of water treatment equipment
 - Drinking water
 - Process water
- Founded in 1965
- Canadian based company
 - Greater Montreal Area



Specialty

- Small systems
 - 15 to 10,000 inhabitants
 - Groundwater
 - Surface Water (Small Flow)

- Contaminant Removal
 - Arsenic
 - Barium
 - Organic carbon
 - Color
 - Hardness
 - Iron
 - Fluoride
 - Manganese
 - Nitrate – Nitrite
 - Dissolved solids
 - Sulfur
 - Turbidity
 - Uranium
 - Others

GUIDELINES



Why We Need to Treat

➤ **Iron:**

- Essential element for human nutrition

Health Effects:

- Gastrointestinal distress

Aesthetic Effects:

- Discoloured water
- Off-flavor (bitter or metallic taste)

➤ **Manganese:**

- Essential element for human nutrition; found in food
- More readily absorbed from drinking water with food

Health Effects:

- Harmful to neurological development in children
- Unproven carcinogen

Aesthetic Effects:

- Discoloured water
- Potential for staining

Guidelines & Treatment Objectives

➤ **Iron:**

MAC: N/A

Aesthetic Objectives:

– Federal:

- 0.3 mg/L (also adopted provincially)
- Newly proposed AO (under consideration): **0.1 mg/L**

➤ **Manganese:**

MAC:

– Federal:

- 0.12 mg/L (also adopted provincially)

Aesthetic Objectives:

– Federal:

- 0.02 mg/L
- Recommended treatment goal: 0.015 mg/L

– Provincial:

- Varies, but most adopting 0.02 mg/L
- Ontario is still officially at 0.05 mg/L.
- BC does not have an AO.

Fe and Mn Removal Treatment Technologies

OVERVIEW



Treatment Technologies

	PROS	CONS
Pyrolusite	<ul style="list-style-type: none">• Meets <0.02 of Mn and <0.3 of Fe even with high concentrations of each	<ul style="list-style-type: none">• Higher backwash flows than Greensand, higher water wastage• No anthracite (more risks of fouling)
Greensand + (CR)	<ul style="list-style-type: none">• Meets <0.02 of MN and <0.3 of Fe even with high concentrations of each• In dept filtration with Anthracite (less fouling risks)• Continuous regeneration of media	<ul style="list-style-type: none">• More chemicals to operate ($\text{KMnO}_4 + \text{Cl}$)



Treatment Technologies

	PROS	CONS
Greensand + (CO)	<ul style="list-style-type: none"> • No KMnO_4 • Easier to operate • Lower backwash flow than Pyrolusite • In dept filtration with Anthracite 	<ul style="list-style-type: none"> • Cannot guarantee lower than 0.05 of Mn • May need to re-condition media with KMnO_4
Biological	<ul style="list-style-type: none"> • No chemicals 	<ul style="list-style-type: none"> • Operational risks • Not suitable for all water qualities • Longer to startup
Ion Exchange Resins (Softeners)	<ul style="list-style-type: none"> • One system for Hardness, Fe and Mn 	<ul style="list-style-type: none"> • No hardness in treated water OR bypass (Fe/Mn residuals) • Not efficient with high concentrations • Some regulatory limitations • Risks of fouling

Treatment Technologies

- Others (rarely used in GW for Fe and Mn)
 - Membrane
 - Needs pre-treatment
 - Can potentially treat colloidal iron

 - Multimedia Filter (Sand Anthracite)
 - Will only eliminate particles of Fe and Mn over 7 microns nominal
 - Will not treat ionised Fe / Mn

 - Sequestration
 - Temporary solution for low levels of Fe and Mn

Problematic #1

ACHIEVING MN <0.02 MG/L

 Problematic #1 – Achieving Mn <math><0.02\text{ mg/l}</math>

What is the problem?

Problematic #1 – Achieving Mn <math>< 0.02\text{ mg/l}</math>

- Built-in resistance to treatment
 - Manganese levels fluctuate from source to source
 - Historical compliance
 - Lack of data
- Some traditional technologies cannot address sufficient removal
 - No previous treatment
 - Sequestrants cannot reduce to 0.02 mg/L since overall Mn content is not reduced

Problematic #1 – Achieving Mn <0.02 mg/l

- Traditional Approaches that will work:
 - Pyrolusite: Can achieve 0.02 mg/L or lower at the cost of higher backwash flows and volume – double than what is required for Greensand
 - Greensand/Greensand Plus: Will effectively reach 0.02 mg/L or less using sodium hypochlorite and potassium permanganate for oxidation and continuous regeneration of the media

Problematic #1 – Achieving Mn <math><0.02\text{ mg/l}</math>

Resistance to Potassium Permanganate

- Additional chemical
- CAPEX, OPEX
- Additional point of maintenance
- Difficult and even unpleasant to manipulate

Problematic #1 – Achieving Mn <0.02 mg/l

Options Moving Forward when using Greensand

- 1. Greensand in catalytic mode (CO)
 - Guaranteed to achieve 0.05 mg/L
 - May achieve 0.02 mg/L but performance may degrade over time
 - Has worked on a pilot scale but long term data at full scale is hard to find.

	Raw Water (Avg)	Treated Water Pilot (Avg)	Treated Water (Current Treatment)
Mn (mg/L)	~0.47	0.000 to 0.007	0.000 to <u>0.036</u>

Problematic #1 – Achieving Mn <0.02 mg/l

Options Moving Forward

- 2. Greensand in CO Mode with re-conditioning
 - Will require periodic media re-conditioning through permanganate soaking.
 - Required frequency for re-conditioning cannot be guaranteed

- 3. Greensand in continuous regeneration mode (CR)
 - Can guarantee performance
 - The media manufacturer will only guarantee 0.05 even with permanganate, but Magnor will guarantee 0.02
 - Making provisions for future use if necessary

 Problematic #1 – Achieving Mn <math><0.02\text{ mg/l}</math>

Biological?

Problematic #2

COLLOIDAL IRON REMOVAL





Problematic #2 – Colloidal Iron

What is the problem?

Problematic #2 – Colloidal Iron

- Particulate iron too small to be filtered
 - Very small
 - Can be smaller than 0.1 microns
 - Causes turbidity
 - Can exceed AO of 0.3 mg/L
 - Could be already oxidized in the water or not

- Could also be organically bound iron hard to oxidize
 - Hypothesis

- Rarely seen in GW
 - But always a risk/uncertainty on Fe removal projects
 - More risk in surface and GUDI water
 - Magnor: 5 cases in 15+ years (4 GW, 1 surface)



Problematic #2 – Colloidal Iron

- 3 pilot studies + 1 at scale situation
 - Summary of all our findings
 - Details and solutions from our last pilot study

Problematic #2 – Colloidal Iron

➤ Summary of what we tried

– Greensand

- Particles too small
- Fe residual too high
- Doesn't work even with Cl + KMnO₄ + Contact Time + pH corrections

– Cartridge filters

- Particles are too small

– Membrane

- Side pilot by others
- Effective but costly
- Other problematics

Problematic #2 – Colloidal Iron

➤ Summary of what we tried

– Ion Exchange

▪ Cationic Ion Exchange Resins (Softener)

- ✓ Works for Ionised Fe
- ✓ Doesn't work for Oxidized Fe
- ✓ Removes all hardness
- ✓ Can foul the resin

▪ Anionic Ion Exchange Resins

- ✓ Performed for 1 day
- ✓ Only works with organically bound iron
- ✓ Would require to regenerate too frequently
- ✓ Long term performance uncertain
- ✓ Can foul the resin

– Magnor's Colloidal Iron Removal System (CIRS)

- In final approval

Problematic #2 – Colloidal Iron

- At Scale Project (Northern Quebec)
 - A well that had no Fe treatment issue
 - Another well was added
 - Fe AO of 0.3mg/L not met
 - After multiple interventions, we realized it probably was colloidal iron

Problematic #2 – Colloidal Iron

- Pilot #4 – Most complete approach
 - Greensand CO Mode
 - Greensand CR Mode
 - pH Corrections
 - Contact Time
 - Ion Exchange Resins

- New approach
 - Along with chlorine and KMnO_4 , add Magnor's CIRS, a new dosing system before the Greensand
 - The objective is to increase the size of particles
 - Promising results

Problematic #2 – Colloidal Iron

Results

- Greensand CO Mode with/without pH correction (current treatment)

	Raw Water (Avg)	Treated Water (Avg)
Fe (mg/L)	0.59	0.26

- Observations
 - ~50% of Iron is removed
 - Lower pH is better, but doesn't solve the problem

Problematic #2 – Colloidal Iron

Results

- Greensand CR Mode without pH correction
 - With KMnO₄ + Chlorine

	Raw Water (Avg)	Treated Water (Avg)
Fe (mg/L)	0.55	0.30

- Observations
 - ~50% of Iron is removed
 - Lower pH is better, but doesn't solve the problem
 - After tests with paper filters, iron is smaller than 0.1 micron

Problematic #2 – Colloidal Iron

Results

- Greensand CR Mode
 - Without pH correction
 - With KMnO₄ + Chlorine
 - Wit 20 min contact time

	Raw Water (Avg)	Treated Water (Avg)
Fe (mg/L)	0.65	0.4

- Observations
 - The contact time doesn't improve the performance

Problematic #2 – Colloidal Iron

Results

- Cationic Ion Exchange Resin (before Greensand)

	Raw Water (Avg)	Treated Water (Avg)
Fe (mg/L)	0.55	0.04

- Observations
 - Iron is removed
 - All hardness is removed, **BUT** not recommended
 - Risk of resin fouling

Problematic #2 – Colloidal Iron

Results

- Cationic Ion Exchange Resin (as polisher after Greensand)

	Fe (Raw Water)	Fe (After GS)	Fe (After Resin)
Sample 1 (mg/L)	N/A	0.51	0.40
Sample 2 (mg/L)	N/A	0.14	0.08
Sample 3 (mg/L)	0.64	0.25	0.19
Sample 4 (mg/L)	0.70	0.10	0.09

- Observations
 - Between 10% and 42% of iron is removed after polishing
 - Iron is oxidized and mainly goes through the resin

Problematic #2 – Colloidal Iron

Results

- Greensand CR Mode + Magnor's **CIRS**
 - With KMnO₄ + Chlorine

	Fe (Raw Water)	Fe (After GS)
Sample 1 (mg/L)	0.76	0.03
Sample 3 (mg/L)	0.74	0.02
Sample 5 (mg/L)	0.75	0.00

- Observations
 - Iron is almost entirely removed using Magnor's CIRS
 - Tried different dosing level, optimization is needed
 - None of the problematics of resins

Problematic #2 – Colloidal Iron

Next Steps

- Get regulatory approval
- Test at full scale
- Test on other water types
 - Surface water
 - Consistent performance on GW

Questions?

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Company Profile

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 - Drinking water
 - Process water
- Founded in 1965
- Canadian based company
 - Greater Montreal Area



Company Profile

- Capacity to handle various project sizes
 - <5 usgpm to >3 000 usgpm
- Complete staff to serve your needs
 - Engineers
 - Designers
 - Project Manager
 - Chemists
 - Service technicians
 - Assemblers-fitters



Specialty

- Small systems
 - 15 to 10,000 inhabitants
 - Groundwater
 - Surface Water (Small Flow)

- Contaminant Removal
 - Arsenic
 - Barium
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 - Turbidity
 - Uranium
 - Others



Market Segments

- Drinking water
 - Municipalities
 - Workers' camp
 - Industries not supplied with water by the municipality
 - Tourism and recreational industry

- Process water
 - Using water in their processes
 - Using water in their product (food)
 - pH neutralization

Our Equipment

- Water Softeners
- Pressure Filters
 - Anthracite
 - Sand
 - Specialized Media
- Activated Carbon Filters
- Greensand Filters
- Ion Exchangers
- Biological filtration



Durable, custom system

- Painted steel tank
- Actuated valve
- Control panel



Fiberglass systems:

- Fiberglass tank
- Integrated control valve

Our Equipment

- Control & Injection Systems
 - Chlorination
 - pH regulation
 - Coagulants

- Ultraviolet Systems

- Membrane Systems

- Preassembled in Container



Peace of Mind

for your water treatment projects

The background of the slide is a light blue water splash with bubbles, creating a clean and fresh aesthetic.

Our Promises

- Long Term Support
- Personalized Service
- Guaranteed Performance



Technical Service

- Local Service Across Canada
 - Start-up and Commissioning
 - Field testing
 - Preventive Maintenance
 - Repair and Refacing
 - Training