City of Toronto, University of Waterloo & Wilfrid Laurier University – Working Together on Corrosion Control

William Fernandes, City of Toronto Dr. Peter Huck, University of Waterloo

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- Serves 3.6 million residents and businesses in Toronto, and portions of York
- Average production: 1,400 ML/d
- Operates and maintains:
 - 4 drinking water treatment plants,
 - 18 pumping stations,
 - 11 underground reservoirs
 - 4 elevated tanks and
 - more than 6,000 km of watermains







Toronto's Lead Problem

- In 2008, Toronto estimated that 65,000 lead service lines (LSLs) existed on City property
- Wide geographic distribution
- No control over lead sources on private property
- >10% of lead samples exceeded the ODWQS of 10 ug/L in 2 of 4 sampling rounds
- Under O.Reg 170/03, Toronto Water was required to prepare a Corrosion Control Plan









Corrosion Control Plan (2010)

- Lead mitigation options identified:
 - 1. Lead source reduction
 - Lead Service Line replacement
 - 2. Upward pH adjustment
 - Add Sodium hydroxide to raise pH to 8.5 or greater
 - **3. Phosphate based inhibition** (zinc phosphate, phosphoric acid, poly-phosphates)
 - Forms a protective coating inside all pipes/fixtures and reduces potential for lead to enter drinking water









Chosen Option for Lead Mitigation

- 3. Phosphate based inhibition using phosphoric acid
 - Regulatory compliance can be achieved quickly
 - Reduces lead in drinking water regardless of lead source
 - LSL replacement to continue
 - Adopted by City Council
 - Supported and endorsed by Toronto Public Health as safe way to reduce lead levels









Corrosion Control Implementation

- Literature review completed on use of phosphate in chloraminated systems
 - Very limited information found
- Testing completed on harvested lead pipe rigs to investigate and confirm:
 - Lead level reductions to be achieved
 - Phosphate dosages required
 - Any potential water quality changes following corrosion control implementation





Corrosion Control Implementation

- Corrosion control systems commissioned at all 4 WTPs over the course of 2014
 - Year 1 and 2: Conditioning dose (3 mg/L as PO₄)
 - Year 3 and onwards: Maintenance dose (1-2 mg/L as PO₄)
- SUCCESS STORY! Significant reduction in lead levels was achieved with implementation of corrosion control treatment using phosphate









Nitrification Event Occurs

- October, 2016: City experienced a nitrification event at one of its reservoirs
- First documented event in Toronto's transmission system
- Sampling confirmed nitrification event in progress with nitrite levels >0.01 mg/L
- Chlorine residual could not be maintained despite every effort taken







What is Nitrification?

- Nitrification is the biologically mediated process of converting ammonia to nitrate
- Can occur in chloraminated systems
- Ammonia is released through chloramine degradation which serves as a nutrient for nitrifying bacteria





What Conditions Promote Nitrification?

Conditions promoting nitrification

- "Low" chlorine residuals (<1 mg/L)
- pH range
- Water temperature >14°C
- High water age: lack of turnover, high hydraulic retention = lower chlorine residuals, higher water temp, increased free ammonia

Presence of nutrients (C:N:P)

Potentially impacted by addition of phosphate for corrosion control





Research Partnership

- Long-term partners: City of Toronto and NSERC Chair led by Dr. Peter Huck at University of Waterloo
- Research team expanded to include microbiology team led by Dr. Robin Slawson at Wilfrid Laurier University
- Research Project: Examine the impacts of phosphoric acid addition for corrosion control on chlorine residual maintenance in the distribution system



UW's NSERC Industrial Research Chair in Water Treatment

- 6th term research focus (2018-2023)
 - Managing raw water quality changes (robustness)
 - Green and innovative treatment processes
 - Distribution system corrosion (lead)
- Multi-disciplinary engineering, microbiology, chemistry
- 9 researchers
- 16 partners (include larger and smaller systems)





Laurier Applied and Environmental Microbiology Lab (Prof. R. M. Slawson)

- Water microbiology focus since 2003
 - Waterborne pathogen persistence
 - Microbial community robustness
 - Factors supporting remedial communities in engineered and natural environments
- Multi-disciplinary research focus (microbiology, biochemistry, aquatic biology) and collaboration (engineering, analytical chemistry)
- Multi-faceted research approach with extensive field sampling experience brought to the bench for both culture-based and molecular assessments



Project Objectives

- Investigate the interrelationships between orthophosphate corrosion inhibitors, chloramine residual, biofilm development, and nitrification in distribution systems
 - Study the effects of different orthophosphate dosages and chloramine initial concentrations
 - Understand the relationship between biofilm development and chloramine decay in the presence of orthophosphate
 - Explore the impact of residence time and orthophosphate presence
 - Study the applicability of key results to other waters (starting soon)



Relationships





Test Procedure

- Two bench-scale systems in parallel (4 months)
 - Control (without OP) and with OP
 - Fed with treated Toronto water prior to disinfection
- pH of the feed water adjusted to 7.4 for all systems



• Temperature: 20°C



FT= feed tank, P= Pump, RT= retention tank, AR= annular reactor, DT= disposal tank

Example Result: Total Chlorine Decay



Error bars included, however they are not always visible due to being smaller than the data point symbol

Conclusions -Physico-Chemical Results

- Free ammonia concentrations and monochloramine decay rates were higher at the higher residence time
- Orthophosphate increased monochloramine decay (more noticeably at the higher residence time)
- Nitrite levels increased at the higher residence time, especially in the presence of orthophosphate, though they remained stable at the lower residence time
- Greatest variation in monochloramine decay, ammonia release and nitrite formation was between residence times rather than orthophosphate presence/absence



Biofilm Formation Potential of Biofilms Removed From Coupons



Conclusions -Microbiological Results

- Microbial biofilm communities monitored every 10-12 days throughout the flow-through experiments
- Microbiological analyses used to create functional and genetic profiling of biofilm community present in annular reactors
- Determined that the addition of orthophosphate impacted microbial growth potential, biofilm formation potential and metabolic diversity
- Results from microbial and physio-chemical analyses may explain changes in disinfectant decay and potential for nitrification



Project Recommendations for Water Utilities

- To reduce the water quality impacts of orthophosphate addition:
 - Reduce orthophosphate dose to the extent possible
 - Increase pH
 - Apply biofilm control strategies
 - Reduce residence time
 - Increase chloramine concentrations at network extremities
 - Increase initial chloramine concentration



What the collaboration has meant for UW and Laurier

- Excellent opportunity for faculty and students involved to experience research as it directly supports real-world problems
- Detailed interactions with Toronto Water staff learning their approach and perspectives



Project Participants

- University of Waterloo
 - Dr. Peter Huck, Dr. Sigrid Peldszus, Mahmoud Badawy, Kimia Aghasadeghi
- Wilfrid Laurier University
 - Dr. Robin Slawson, Mitchell Cooke, Vedika Bakshi
- Toronto Water
 - Liza Ballantyne, Gary Che, Alireza Anvari, Sarah Wilson





Thank you!

William Fernandes, City of Toronto <u>William.Fernandes@Toronto.ca</u>

Dr. Peter Huck, University of Waterloo pmhuck2@uwaterloo.ca



