

Legacy Aluminum in the Distribution System: Impacts on Corrosion Control with Orthophosphate

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Presentation Outline

- Legacy aluminum in the distribution system
 - How does it get there?
 - What are the impacts?
- Implementing orthophosphate in the presence of legacy aluminum: case studies
 - Selecting the appropriate orthophosphate dosage & developing an implementation plan
 - Observed lead reduction at 2 mg/L as PO₄
 - Observations of secondary impacts and their control
- Summary

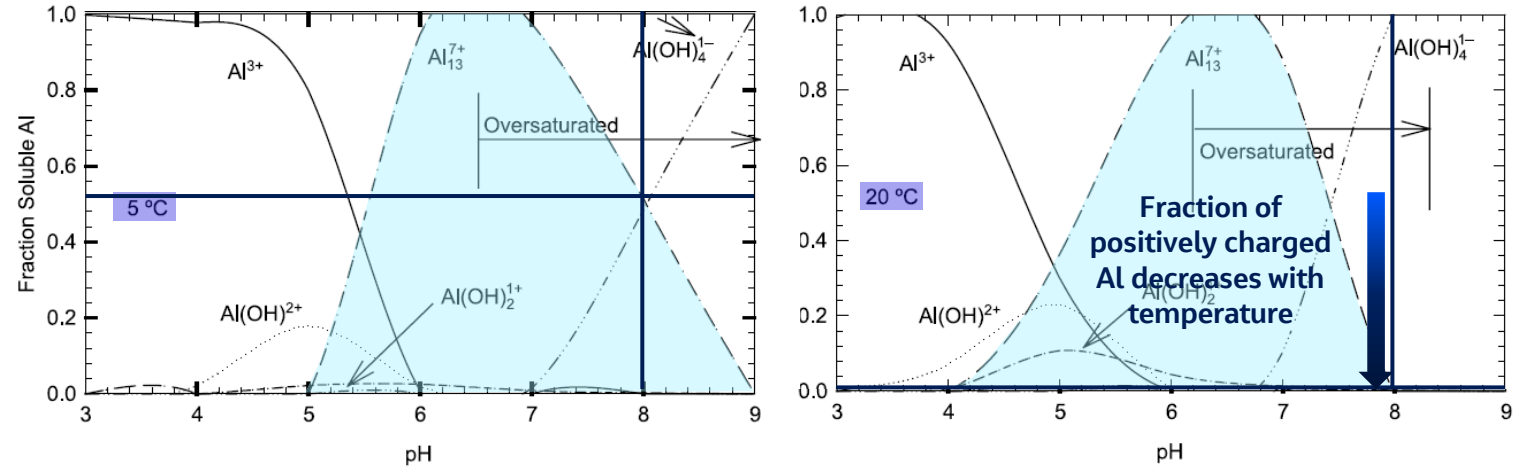
Legacy Aluminum in the Distribution System

Sources and Impacts

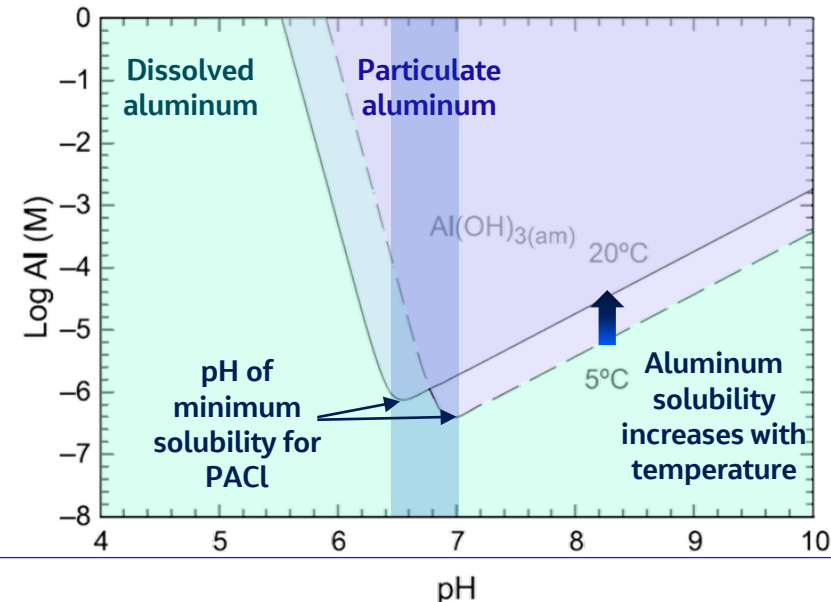
How Does Aluminum Get in the Distribution System?

- Aluminum can be present in treated water from the use of **aluminum-based coagulants** in a **non-optimized coagulation** process
- The **speciation** and **solubility** of coagulants are impacted by **temperature and pH**, which vary seasonally
- Many WTPs treating Great Lakes waters experience **high residual dissolved aluminum in treated water**, particularly in the **summer months**
- Aluminum can also be released in localized areas from **cementitious pipes** and **cement mortar linings** when watermains are relined

Theoretical aluminum speciation for PACl, $Al_T = 1 \text{ mg/L}$. Adapted from Pernitsky and Edzwald, 2006



Theoretical solubility diagram for PACl in deionized water. Adapted from Pernitsky and Edzwald, 2006



What Happens to Dissolved Aluminum in the Distribution System?

- **Post-precipitation** of dissolved aluminum and **flocculation** can occur in the distribution system if the pH decreases below the pH of coagulation
- Deposition of **gelatinous aluminum-containing substances** or **aluminum-containing crystalline solids** in watermains may also occur
- **Accumulation** can occur in areas of low flow, such as dead-ends and reservoirs
- **Release** of deposits at **very high concentration** can occur from hydraulic surges such as watermain breaks



Example of gelatinous aluminum deposit



Example of aluminum silicate scale



Flush water containing 38 mg/L aluminum

What Are the Impacts of Aluminum in the Distribution System?

Hydraulic/ Pressure

- **Increased hydraulic friction from deposits/scales**, reducing the distribution system's carrying capacity, increasing pressure losses, and increasing the energy required for pumping
- Deposition within water meters and service lines can cause **water meter malfunctions and low household water pressure**

Water Quality

- Floc and released deposits can **impart turbidity and/or milky colour** to the water, causing **aesthetic impairment**
- Floc, deposits, and scales can **provide a substrate upon which other trace metals can adsorb or co-precipitate**
- Accumulated trace metals can **be released at high concentrations** due to desorption or particle detachment
- **Dissolution of previously precipitated deposits and scales** can be a source of aluminum

Secondary Disinfection

- High levels of aluminum may interfere with secondary disinfection due to **enmeshment and protection of microorganisms**

Corrosion Control

- The reaction of orthoPO₄ with aluminum produces a **precipitate that can give water a “milky” appearance**
- Aluminum can behave as a sink for dissolved orthoPO₄, **reducing the residual available for corrosion control**
- Deposits can lessen the efficacy of corrosion control by **interfering with the development of protective corrosion scales**
- Impacts expected with orthoPO₄ >3.0 mg/L as PO₄ and aluminum >0.1 mg/L

Guidelines for Aluminum in Treated Water and Distribution System

Guideline	Value	Applicability	Purpose
USEPA Secondary MCL	50 to 200 µg/L	Applies to point-of-entry	Prevent discoloured water
Operational Guideline (Ontario) <i>Technical Support Document for Ontario Drinking Water Standards, Objectives and Guidelines (2003, rev. 2006)</i>	100 µg/L	Applies to point-of-entry	Prevent coating of pipes in the distribution system, interferences with certain industrial processes, and flocculation in the distribution system.
Operational Guideline (Health Canada) <i>Guidelines for Canadian Drinking Water Quality: Guideline Technical Document – Aluminum (2021)</i>	100 µg/L	Applies to point-of-entry and distribution system . For facilities that use aluminum-based coagulants, the OG applies to a locational running annual average calculated based on daily samples that are averaged on a monthly basis.	Minimize accumulation and release of aluminum and co-occurring contaminants and coating of pipes and appurtenances.
Maximum Acceptable Concentration (Health Canada) <i>Guidelines for Canadian Drinking Water Quality: Guideline Technical Document – Aluminum (2021)</i>	2,900 µg/L	Applies to the distribution system and point-of-consumption . For facilities that use aluminum-based coagulants, the MAC applies to a locational running annual average calculated based on a minimum of monthly samples collected from the distribution system.	Health-based (neurological effects observed in rats)

Consider a target <100 µg/L when implementing phosphate-based corrosion control

Implementing Corrosion Control in Systems With Legacy Aluminum

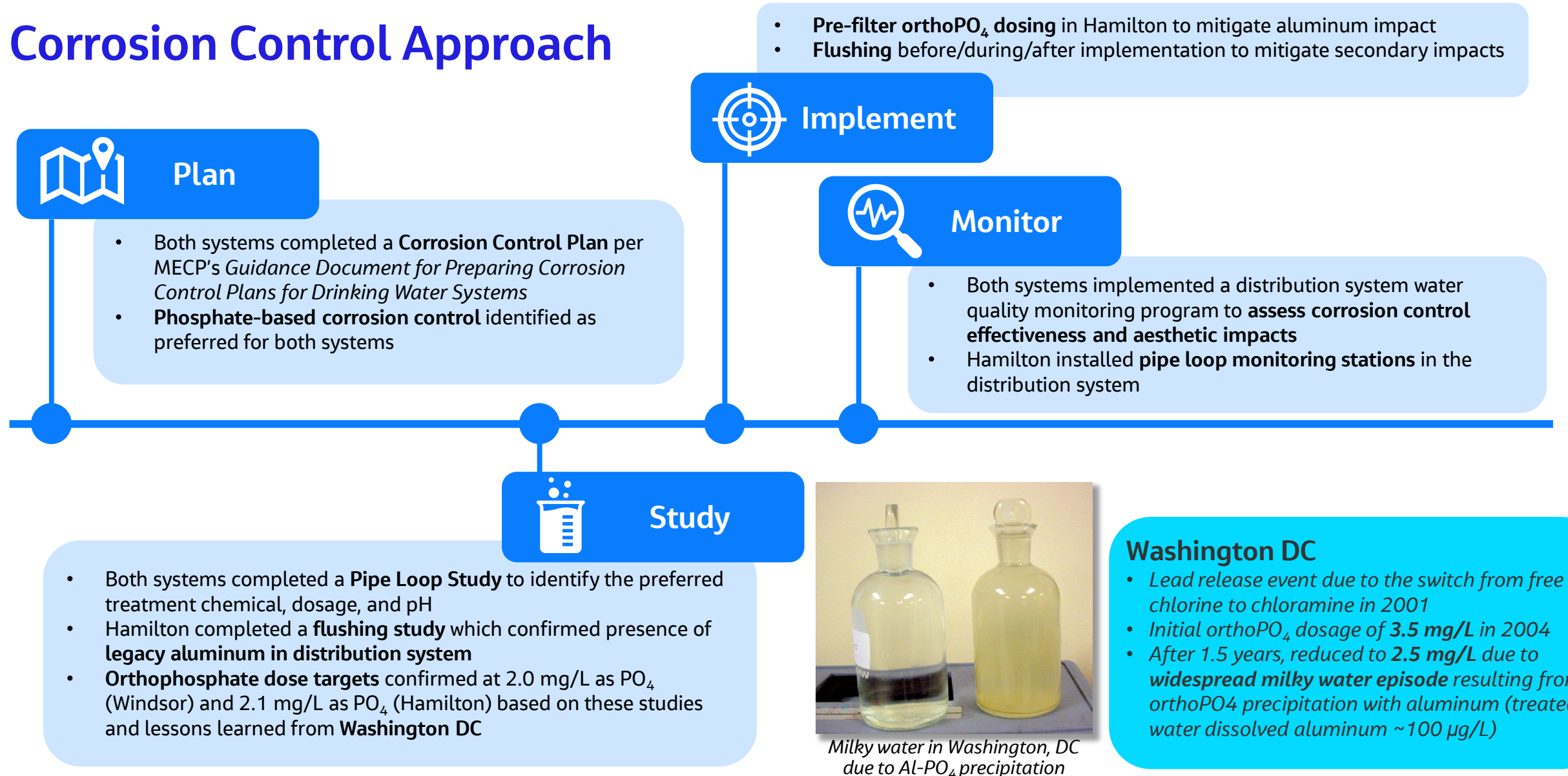
Strategies and Outcomes

System Characteristics



	Windsor	Hamilton
Serviced population	~270,000	~570,000
Size of distribution system	~1,100 km	~2,000 km
Source water	Detroit River (Lake St. Clair)	Lake Ontario
pH control	CO ₂ ahead of coagulation	None
Corrosion control treatment	Orthophosphate, 2.0 mg/L as PO ₄	Orthophosphate, 2.1 mg/L as PO ₄
Year implemented	August 2016	November 2018
Average treated water pH	7.0	7.8
Average treated water alkalinity	86 mg/L as CaCO ₃	86 mg/L as CaCO ₃
Secondary disinfectant	Free chlorine	Chloramine

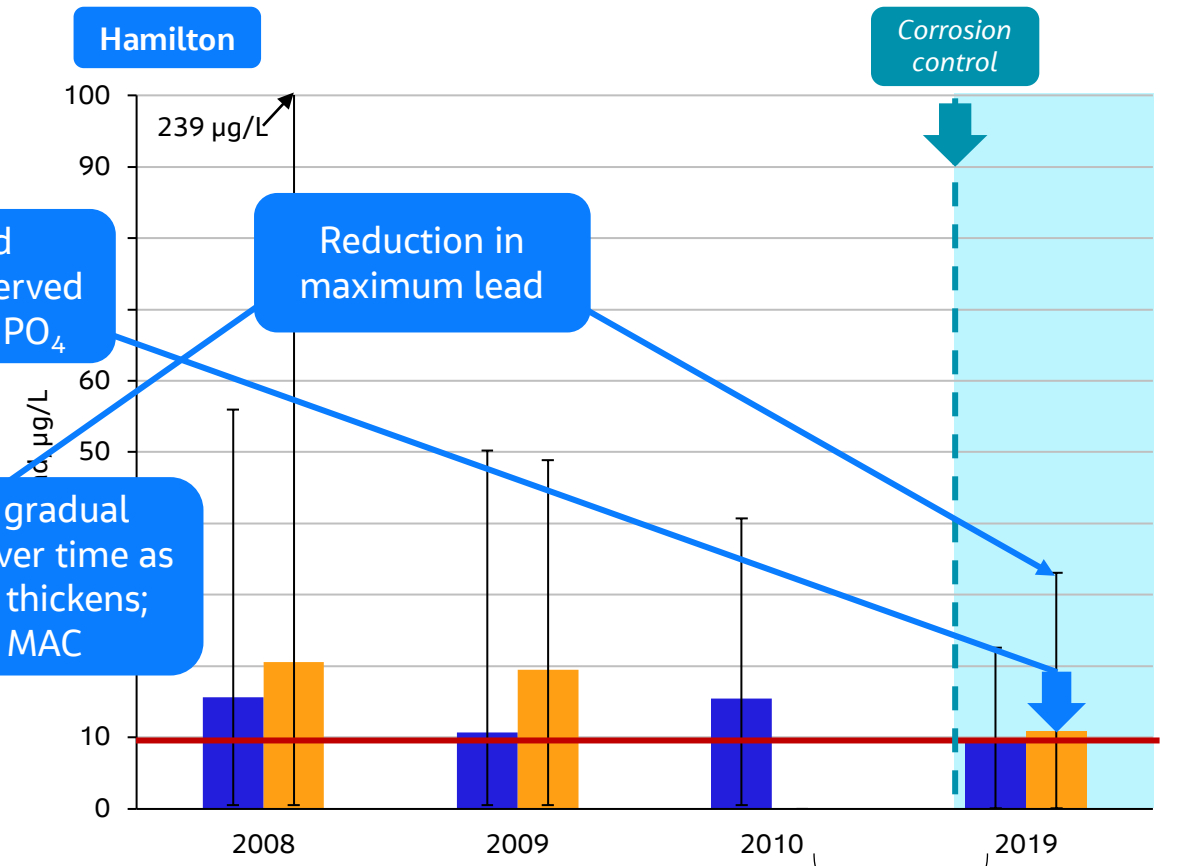
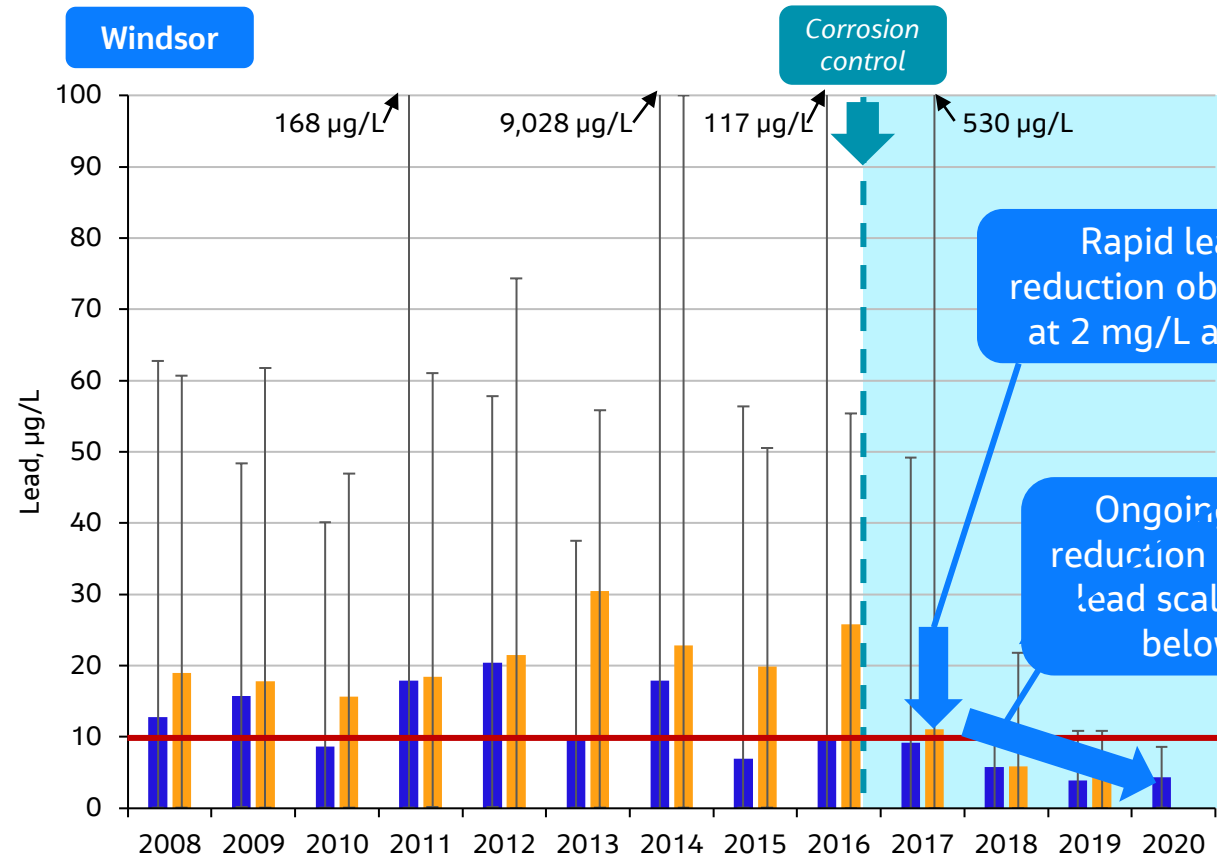
Corrosion Control Approach



Observed Lead Reduction – At Tap



Schedule 15.1 Residential and Non-Residential Lead, 90th Percentiles

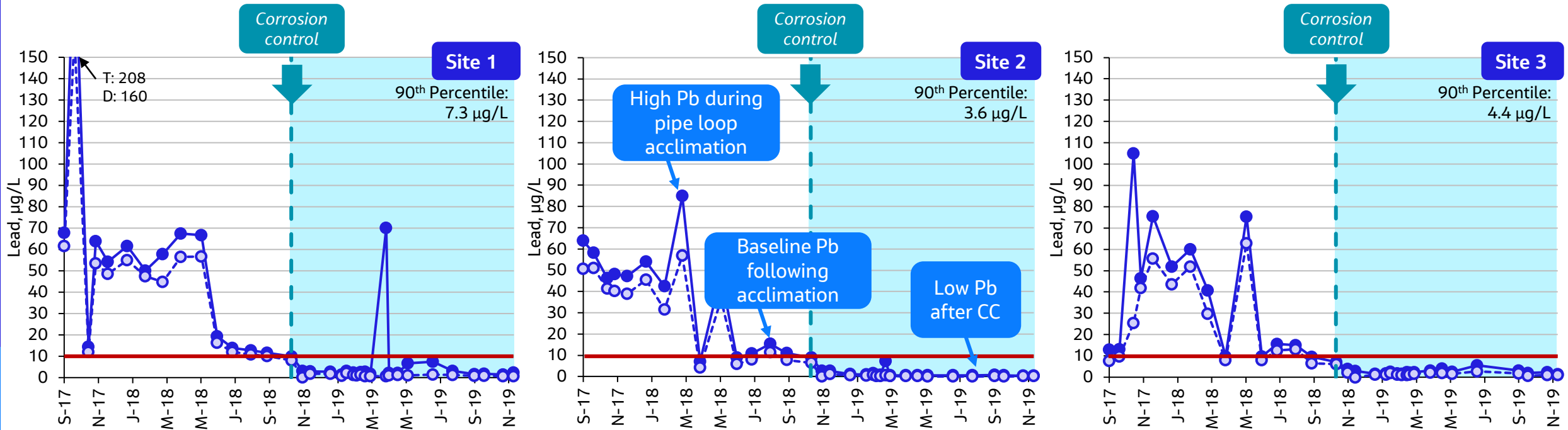


Regulatory relief from at-tap sampling

Observed Lead Reduction – Pipe Loop Monitoring Stations

- Total lead
- Dissolved lead
- MAC

Hamilton



Pipe loop monitoring stations confirmed results observed from at-tap sampling
 Post-corrosion-control lead consistently below MAC

Evidence of Lead-Phosphate Scale Formation

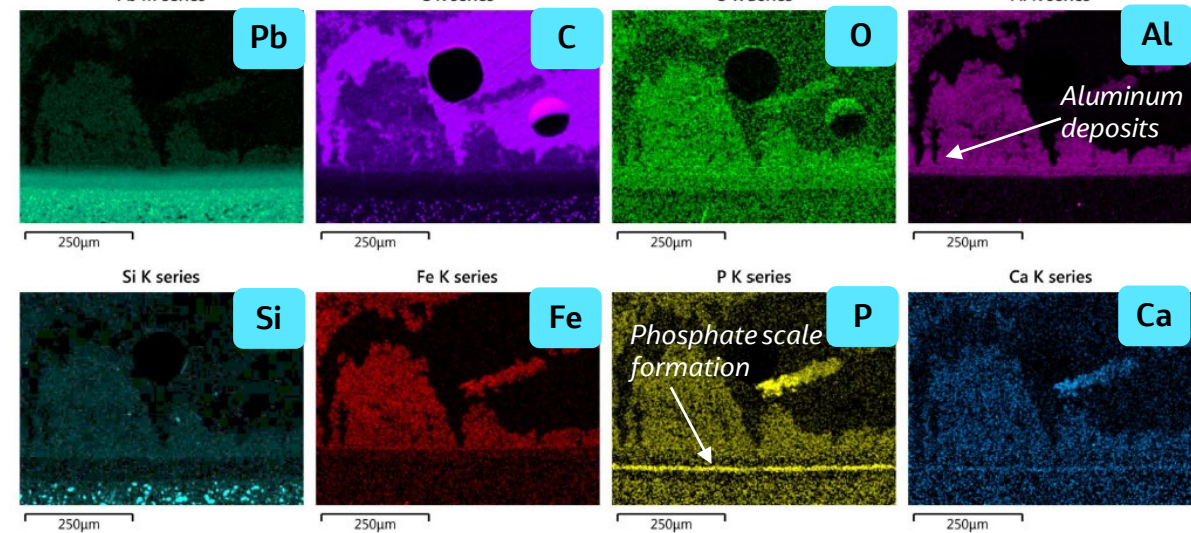
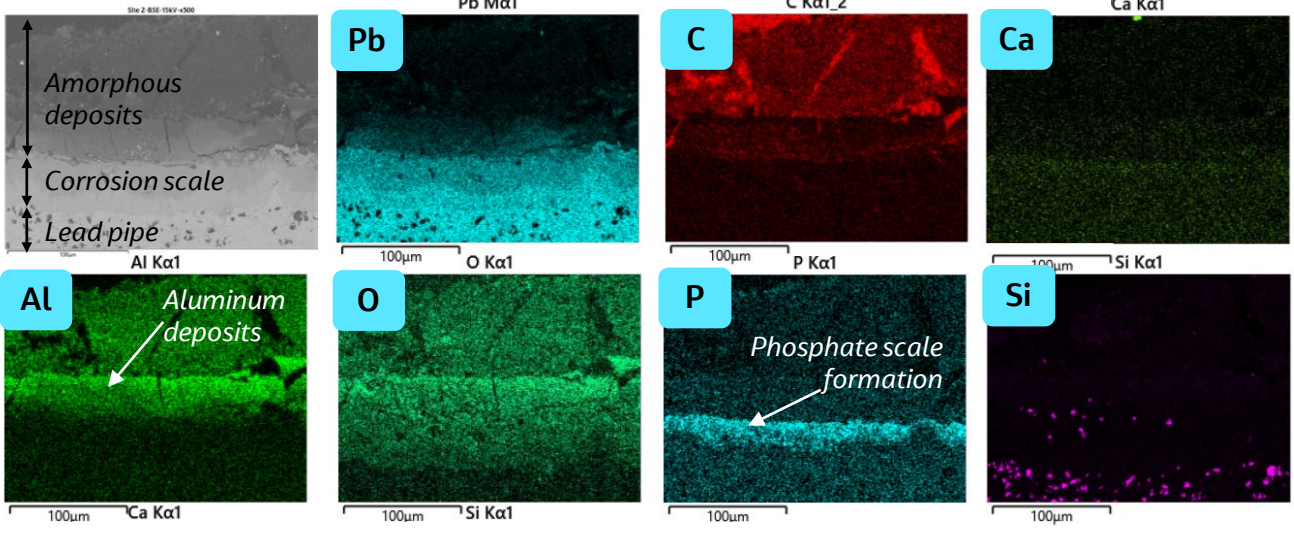
OrthoPO₄ has penetrated amorphous aluminum deposits to form phosphate-based lead scale at the pipe surface

SEM/EDS and XRD Scale Analysis

Windsor



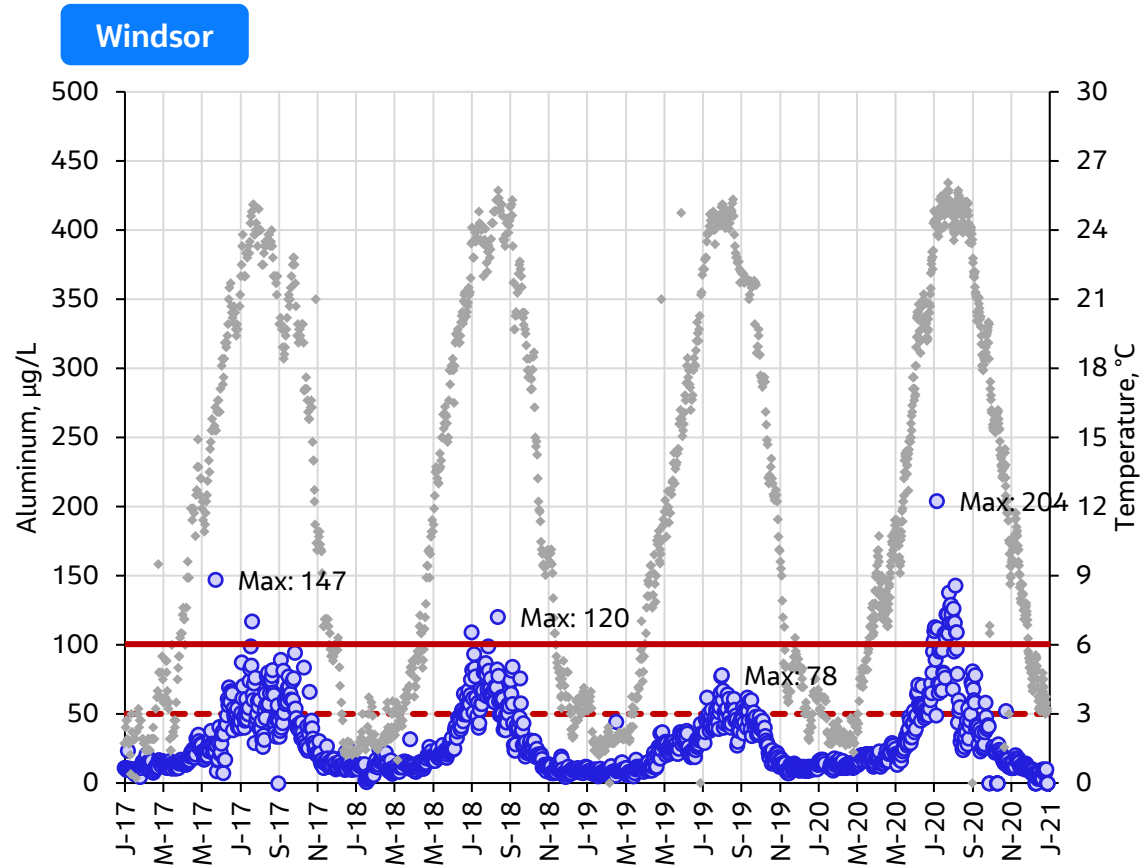
Hamilton



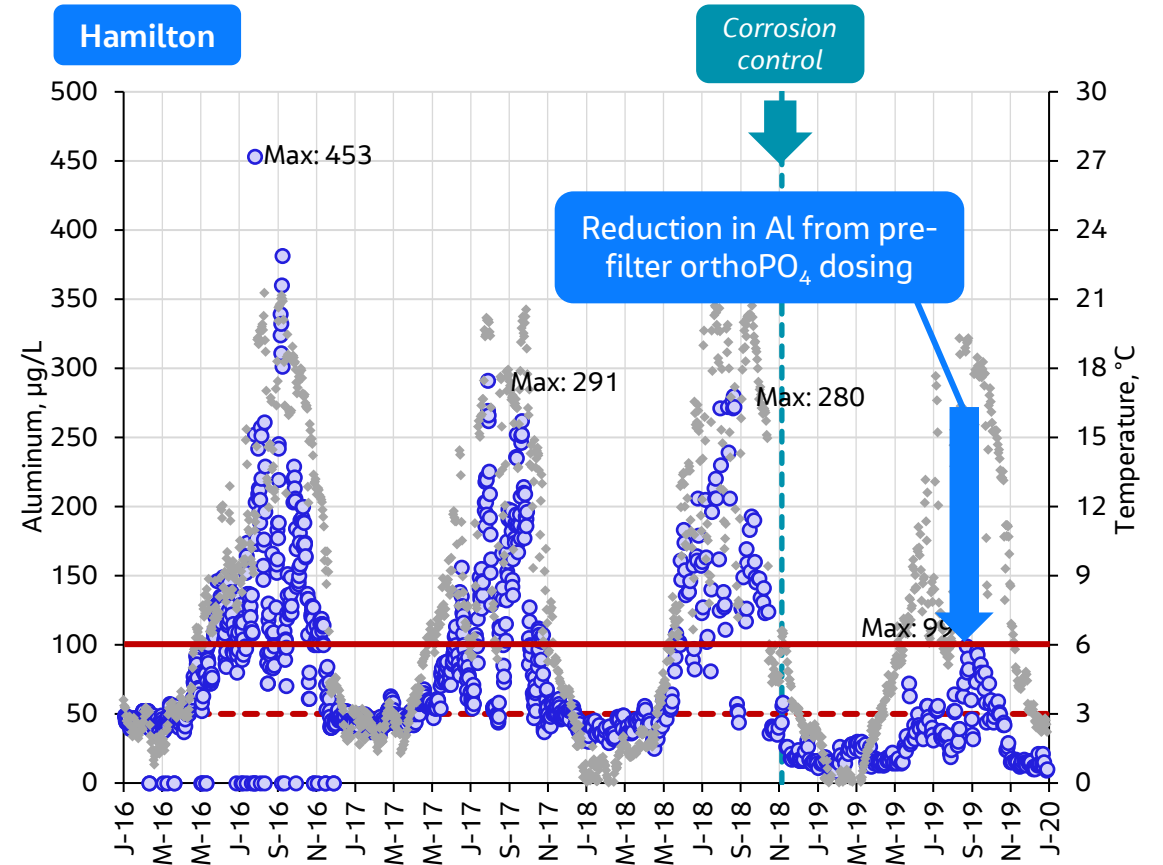
SEM/EDS = Scanning Electron Microscopy / Energy Dispersive X-Ray Spectroscopy; XRD = X-Ray Diffraction

Residual Aluminum in Treated Water

- Treated water aluminum
- ◆ Raw water temperature
- - - Aluminum Best Practice (50 µg/L)
- Aluminum Operational Guideline (100 µg/L)



pH control with CO₂ (target 7.1)
 Summer Al peaks above OG
 Annual avg. Al below best practice

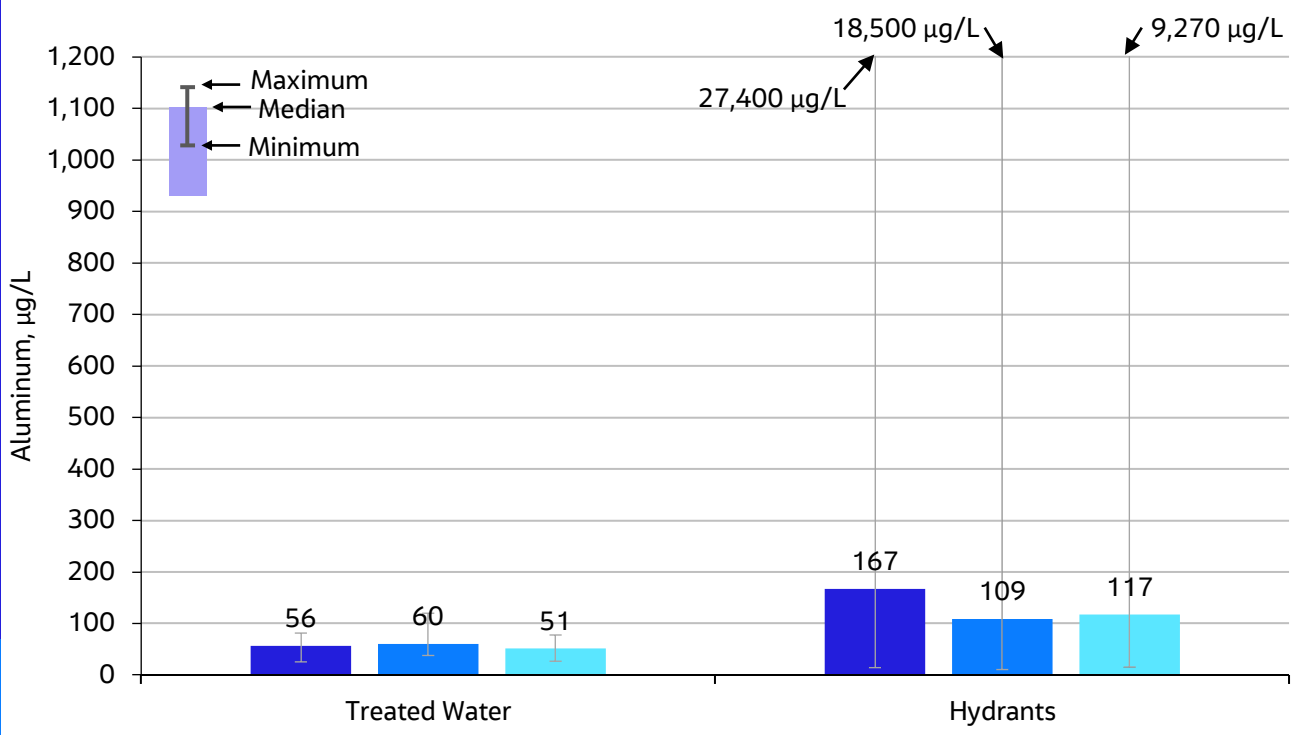


No pH control
Pre-CC: Summer Al peaks well above OG, annual avg. Al above best practice but below OG
Post-CC: Summer Al peaks around OG, annual avg. Al below best practice

Aluminum in Hydrants and Residential Samples – Windsor

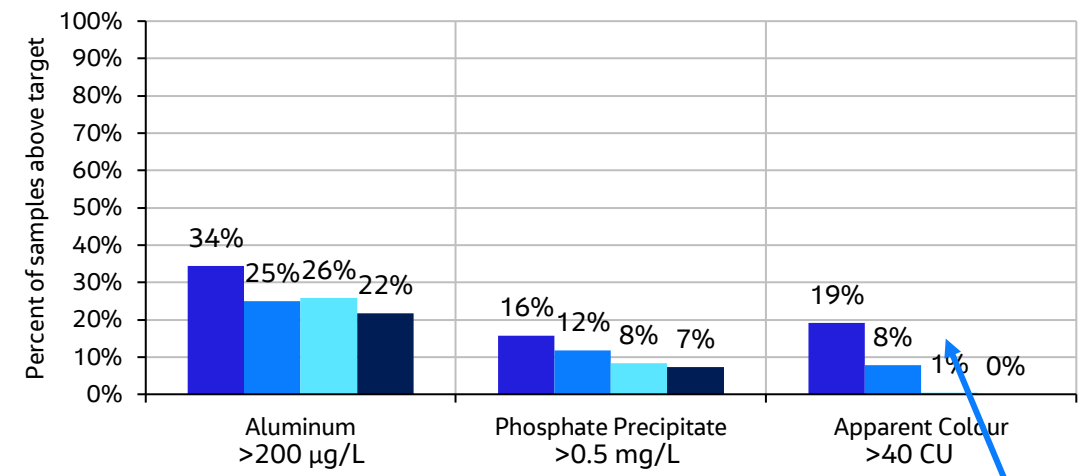


Median, Min, and Max Aluminum – Summer Sampling Rounds

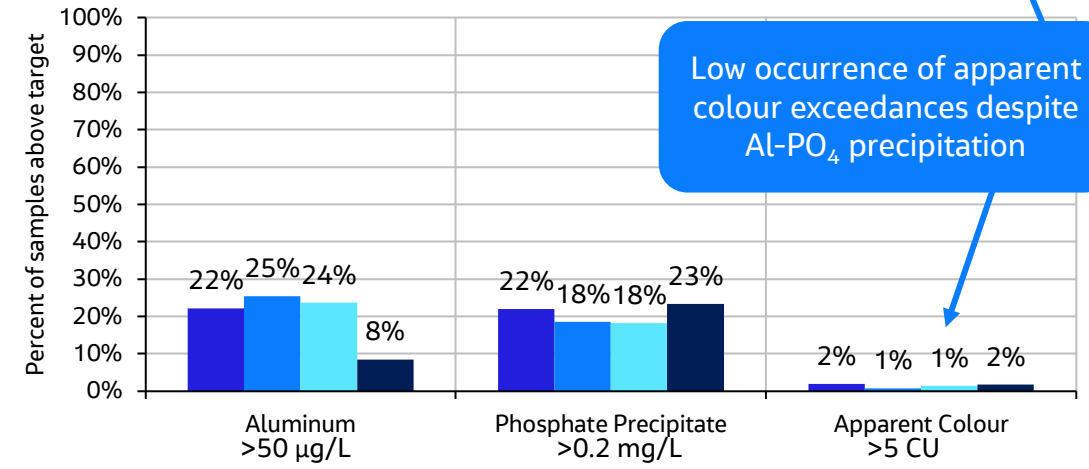


Significant accumulation and release of aluminum can occur even with treated water aluminum below OG – important to monitor
 Aluminum-phosphate precipitation was observed but did not result in widespread aesthetic impact like that observed in Washington DC

Percent of Samples Above Target – Hydrants



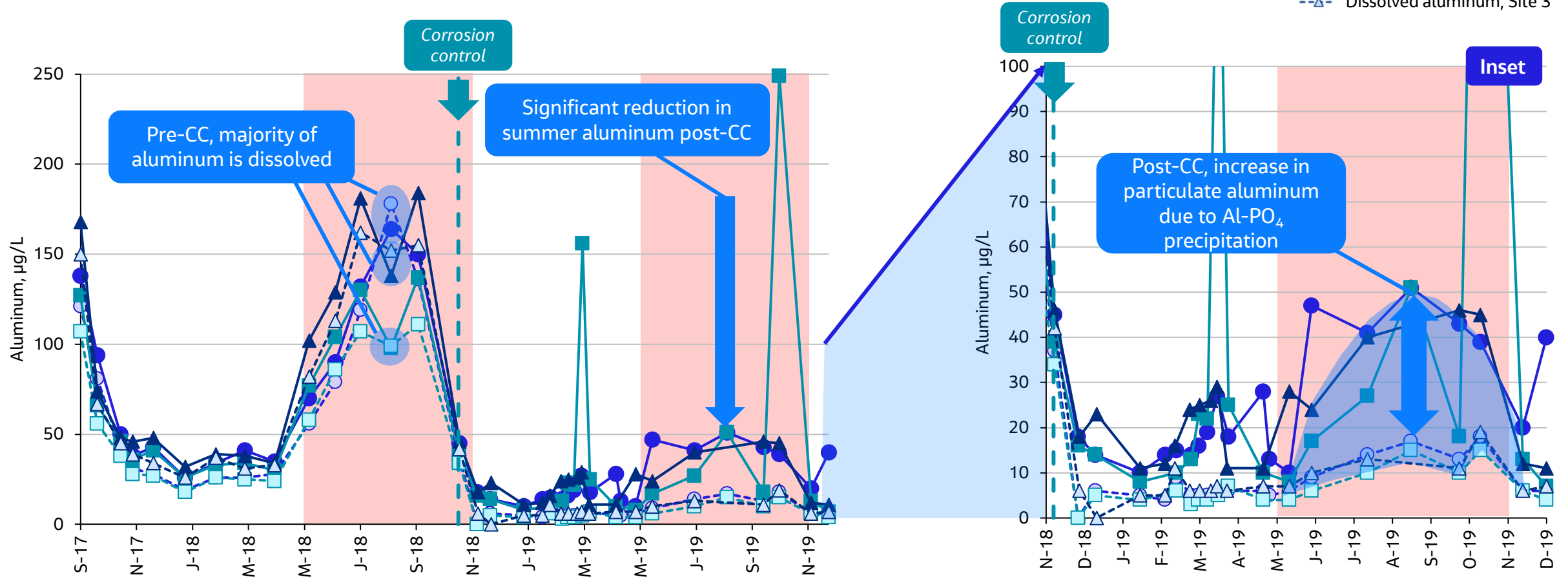
Percent of Samples Above Target – Residential Samples



Low occurrence of apparent colour exceedances despite Al-PO₄ precipitation

Aluminum in Pipe Loop Monitoring Stations – Hamilton

- Total aluminum, Site 1
- Total aluminum, Site 2
- ▲ Total aluminum, Site 3
- Dissolved aluminum, Site 1
- Dissolved aluminum, Site 2
- △ Dissolved aluminum, Site 3



Prior to corrosion control, majority of aluminum in the distribution system was dissolved
 Reduction in treated water Al from pre-filter orthoPO₄ dosing has helped control Al in the distribution system to ~50 µg/L during warmer months
 Al precipitation is occurring and has increased post-corrosion-control and the precipitate is mobile

Proactive Flushing

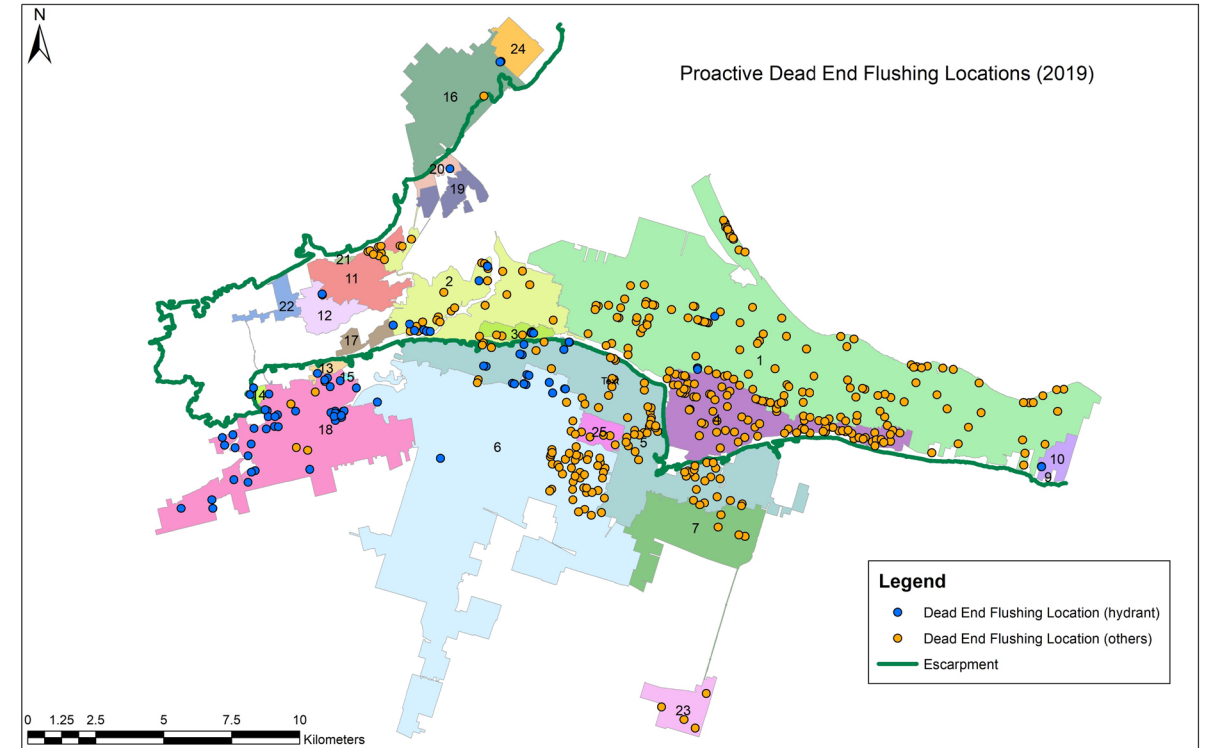
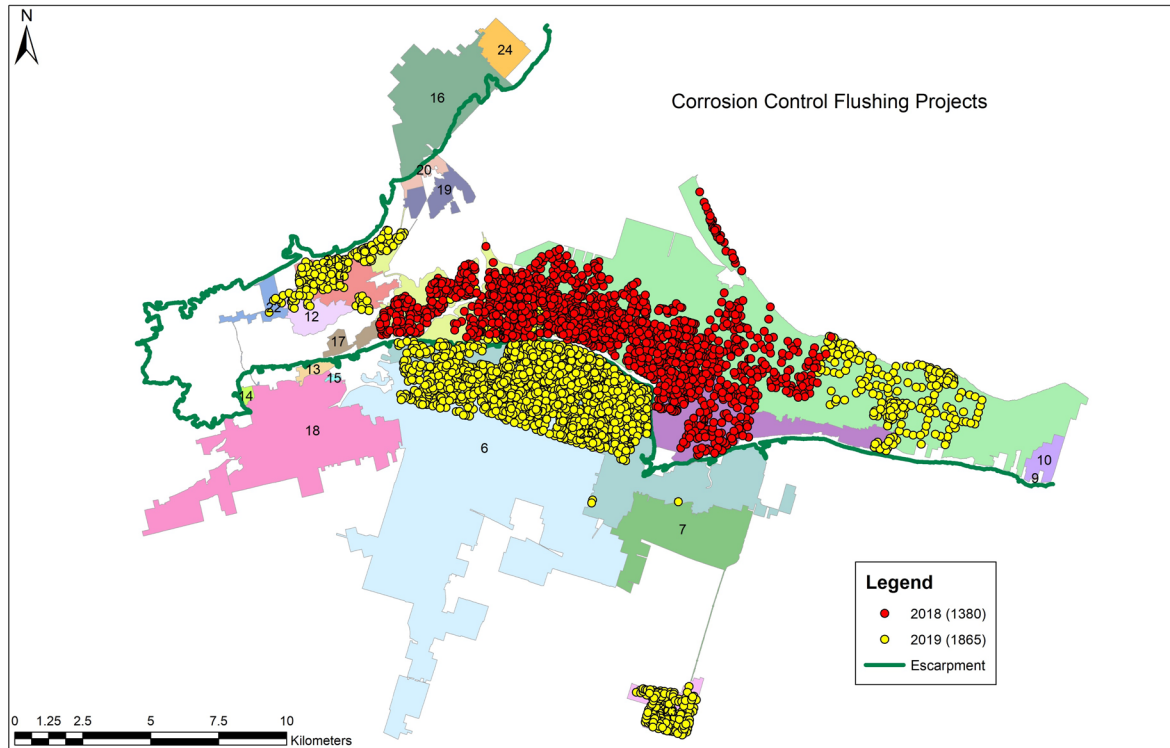
Hamilton

Windsor

Dead-end flushing
Conventional flushing (fire-flow testing)

Hamilton

Dead-end flushing
Conventional flushing (fire-flow testing)
UDF + quasi-UDF (corrosion control)



Flushing has helped to remove accumulated aluminum, spread orthoPO₄, and maintain chlorine residual by temporarily reducing water age

Summary

Key Takeaways

Summary

- **Residual aluminum** in treated water from the use of **aluminum-based coagulants** can **accumulate in distribution systems**, potentially impacting hydraulics, water quality, secondary disinfection, & corrosion control
- **Coagulation optimization** can reduce residual dissolved aluminum in treated water
- Two systems with **legacy aluminum** implemented **phosphate-based corrosion control**
 - An initial orthophosphate dose of **2 mg/L as PO₄** was applied in anticipation of potential **aesthetic impacts** caused by the precipitation of aluminum with orthoPO₄ above 3 mg/L as PO₄
 - In the absence of pH control at their WTP, Hamilton utilized **pre-filter orthoPO₄ dosing** to precipitate and remove aluminum in the treatment process, **preventing it from entering the distribution system**
 - **Rapid lead reduction** was observed at 2 mg/L as PO₄ with gradual reduction ongoing
 - **Aluminum-based amorphous deposits** were confirmed to be present in lead service lines, but scale analysis confirmed that orthoPO₄ had penetrated these amorphous deposits, forming **phosphate-based scale at the pipe surface**
 - Despite evidence of aluminum-phosphate precipitation, **no widespread “milky water” aesthetic impact** was observed in either system, due to the lower initial orthophosphate dose and control of treated water aluminum to <100 µg/L
 - **Distribution system water quality monitoring** and **flushing** were key strategies used to proactively manage aluminum-phosphate precipitation

Acknowledgements

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Thank You!

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